

## OCTANOL-WATER PARTITION COEFFICIENTS

The octanol-water partition coefficient,  $P$ , is a widely used parameter for correlating biological effects of organic substances. It is a property of the two-phase system in which water and 1-octanol are in equilibrium at a fixed temperature and the substance is distributed between the water-rich and octanol-rich phases.  $P$  is defined as the ratio of the equilibrium concentration of the substance in the octanol-rich phase to that in the water-rich phase, in the limit of zero concentration. In general,  $P$  tends to be large for compounds with extended non-polar structures (such as long chain or multi-ring hydrocarbons) and small for compounds with highly polar groups. Thus  $P$  (or, in its more common form of expression,  $\log P$ ) provides a measure of the lipophilic vs. hydrophilic nature of a compound, which is an important consideration in assessing the potential toxicity. A discussion of methods of measurement and accuracy considerations for  $\log P$  may be found in Reference 1.

This table gives selected values of  $\log P$  for about 450 organic compounds, including many of environmental importance. All values refer to a nominal temperature of 25 °C. The source of each value is indicated in the last column. These references contain data on many more compounds than are included here.

Mol. form.	Name	$\log P$	Ref.
CCl <sub>2</sub> F <sub>2</sub>	Dichlorodifluoromethane	2.16	2
CCl <sub>3</sub> F	Trichlorofluoromethane	2.53	2
CCl <sub>4</sub>	Tetrachloromethane	2.64	2
CHBr <sub>3</sub>	Tribromomethane	2.38	2
CHCl <sub>3</sub>	Trichloromethane	1.97	2
CH <sub>2</sub> BrCl	Bromochloromethane	1.41	2
CH <sub>2</sub> Br <sub>2</sub>	Dibromomethane	2.3	2
CH <sub>2</sub> Cl <sub>2</sub>	Dichloromethane	1.25	2
CH <sub>2</sub> F <sub>2</sub>	Difluoromethane	0.20	1
CH <sub>2</sub> I <sub>2</sub>	Diiodomethane	2.5	2
CH <sub>2</sub> O	Formaldehyde	0.35	1
CH <sub>2</sub> O <sub>2</sub>	Formic acid	-0.54	1
CH <sub>3</sub> Br	Bromomethane	1.19	2
CH <sub>3</sub> Cl	Chloromethane	0.91	2
CH <sub>3</sub> F	Fluoromethane	0.51	1
CH <sub>3</sub> I	Iodomethane	1.5	2
CH <sub>3</sub> NO	Formamide	-1.51	1
CH <sub>3</sub> NO <sub>2</sub>	Nitromethane	-0.33	1
CH <sub>4</sub> O	Methanol	-0.74	1
CH <sub>5</sub> N	Methylamine	-0.57	1
C <sub>2</sub> Cl <sub>3</sub> F <sub>3</sub>	1,1,2-Trichlorotrifluoroethane	3.16	2
C <sub>2</sub> Cl <sub>4</sub>	Tetrachloroethylene	2.88	2
C <sub>2</sub> Cl <sub>6</sub>	Hexachloroethane	4.00	4
C <sub>2</sub> HCl <sub>3</sub>	Trichloroethylene	2.53	2
C <sub>2</sub> HCl <sub>5</sub>	Pentachloroethane	2.89	2
C <sub>2</sub> H <sub>2</sub> Cl <sub>2</sub>	1,1-Dichloroethylene	2.13	2
C <sub>2</sub> H <sub>2</sub> Cl <sub>2</sub>	<i>cis</i> -1,2-Dichloroethylene	1.86	2
C <sub>2</sub> H <sub>2</sub> Cl <sub>2</sub>	<i>trans</i> -1,2-Dichloroethylene	1.93	2
C <sub>2</sub> H <sub>2</sub> Cl <sub>4</sub>	1,1,2,2-Tetrachloroethane	2.39	2
C <sub>2</sub> H <sub>3</sub> Cl	Chloroethylene	1.38	2
C <sub>2</sub> H <sub>3</sub> Cl <sub>3</sub>	1,1,1-Trichloroethane	2.49	2
C <sub>2</sub> H <sub>3</sub> Cl <sub>3</sub>	1,1,2-Trichloroethane	2.38	2
C <sub>2</sub> H <sub>3</sub> N	Acetonitrile	-0.34	1
C <sub>2</sub> H <sub>4</sub> Cl <sub>2</sub>	1,1-Dichloroethane	1.79	2
C <sub>2</sub> H <sub>4</sub> Cl <sub>2</sub>	1,2-Dichloroethane	1.48	2
C <sub>2</sub> H <sub>4</sub> O	Acetaldehyde	0.45	1
C <sub>2</sub> H <sub>4</sub> O	Ethylene oxide	-0.30	1

Compounds are listed by molecular formula following the Hill convention. To locate a compound by name or CAS Registry Number when the molecular formula is not known, use the table "Physical Constants of Organic Compounds" in Section 3 and its indexes to determine the molecular formula.

### References

1. Sangster, J., *J. Phys. Chem. Ref. Data*, 18, 1111, 1989.
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3. Shiu, W. Y., and Mackay, D., *J. Phys. Chem. Ref. Data*, 15, 911, 1986.
4. Pinsuwan, S., Li, L., and Yalkowsky, S. H., *J. Chem. Eng. Data*, 40, 623, 1995.
5. *Solubility Data Series*, International Union of Pure and Applied Chemistry, Vol. 20, Pergamon Press, Oxford, 1985.
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Mol. form.	Name	$\log P$	Ref.
C <sub>2</sub> H <sub>4</sub> O <sub>2</sub>	Acetic acid	-0.17	1
C <sub>2</sub> H <sub>5</sub> Br	Bromoethane	1.6	2
C <sub>2</sub> H <sub>5</sub> Cl	Chloroethane	1.43	2
C <sub>2</sub> H <sub>5</sub> I	Iodoethane	2	2
C <sub>2</sub> H <sub>5</sub> NO	Acetamide	-1.26	1
C <sub>2</sub> H <sub>5</sub> NO <sub>2</sub>	Nitroethane	0.18	1
C <sub>2</sub> H <sub>6</sub> O	Ethanol	-0.30	1
C <sub>2</sub> H <sub>6</sub> O	Dimethyl ether	0.10	1
C <sub>2</sub> H <sub>6</sub> OS	Dimethyl sulfoxide	-1.35	1
C <sub>2</sub> H <sub>6</sub> O <sub>2</sub> S	Dimethyl sulfone	-1.41	1
C <sub>2</sub> H <sub>7</sub> N	Ethylamine	-0.13	1
C <sub>2</sub> H <sub>7</sub> N	Dimethylamine	-0.38	1
C <sub>3</sub> H <sub>3</sub> N	2-Propenenitrile	0.25	1
C <sub>3</sub> H <sub>3</sub> Cl <sub>2</sub>	<i>cis</i> -1,3-Dichloropropene	2.03	2
C <sub>3</sub> H <sub>3</sub> O	Propargyl alcohol	-0.38	1
C <sub>3</sub> H <sub>4</sub> O	Acrolein	-0.01	1
C <sub>3</sub> H <sub>5</sub> Br	3-Bromopropene	1.79	1
C <sub>3</sub> H <sub>5</sub> ClO	Epichlorohydrin	0.30	2
C <sub>3</sub> H <sub>5</sub> Cl <sub>3</sub>	1,2,3-Trichloropropane	2.63	2
C <sub>3</sub> H <sub>5</sub> N	Propanenitrile	0.16	1
C <sub>3</sub> H <sub>5</sub> NO	Acrylamide	-0.78	1
C <sub>3</sub> H <sub>6</sub> Cl <sub>2</sub>	1,2-Dichloropropane	2.0	2
C <sub>3</sub> H <sub>6</sub> O	Allyl alcohol	0.17	1
C <sub>3</sub> H <sub>6</sub> O	Propanal	0.59	1
C <sub>3</sub> H <sub>6</sub> O	Acetone	-0.24	1
C <sub>3</sub> H <sub>6</sub> O	Methyloxirane	0.03	1
C <sub>3</sub> H <sub>6</sub> O <sub>2</sub>	Propanoic acid	0.33	1
C <sub>3</sub> H <sub>6</sub> O <sub>2</sub>	Methyl acetate	0.18	1
C <sub>3</sub> H <sub>7</sub> Br	1-Bromopropane	2.1	2
C <sub>3</sub> H <sub>7</sub> Br	2-Bromopropane	1.9	2
C <sub>3</sub> H <sub>7</sub> Cl	1-Chloropropane	2.04	1
C <sub>3</sub> H <sub>7</sub> Cl	2-Chloropropane	1.90	1
C <sub>3</sub> H <sub>7</sub> I	1-Iodopropane	2.5	2
C <sub>3</sub> H <sub>7</sub> N	Allylamine	0.03	1
C <sub>3</sub> H <sub>7</sub> NO	<i>N,N</i> -Dimethylformamide	-1.01	1
C <sub>3</sub> H <sub>7</sub> NO	<i>N</i> -Methylacetamide	-1.05	1
C <sub>3</sub> H <sub>7</sub> NO <sub>2</sub>	1-Nitropropane	0.87	1

Mol. form.	Name	log <i>P</i>	Ref.	Mol. form.	Name	log <i>P</i>	Ref.
C <sub>3</sub> H <sub>8</sub> O	1-Propanol	0.25	1	C <sub>5</sub> H <sub>10</sub> O	2-Methyltetrahydrofuran	1.85	2
C <sub>3</sub> H <sub>8</sub> O	2-Propanol	0.05	1	C <sub>5</sub> H <sub>10</sub> O <sub>2</sub>	Pentanoic acid	1.39	1
C <sub>3</sub> H <sub>8</sub> S	1-Propanethiol	1.81	1	C <sub>5</sub> H <sub>10</sub> O <sub>2</sub>	Propyl acetate	1.24	1
C <sub>3</sub> H <sub>9</sub> N	Propylamine	0.48	1	C <sub>5</sub> H <sub>10</sub> O <sub>2</sub>	Ethyl propanoate	1.21	1
C <sub>3</sub> H <sub>9</sub> N	Isopropylamine	0.26	1	C <sub>5</sub> H <sub>10</sub> O <sub>3</sub>	Diethyl carbonate	1.21	1
C <sub>3</sub> H <sub>9</sub> N	Ethylmethylamine	0.15	1	C <sub>5</sub> H <sub>11</sub> Br	1-Bromopentane	3.37	1
C <sub>3</sub> H <sub>9</sub> N	Trimethylamine	0.16	1	C <sub>5</sub> H <sub>11</sub> F	1-Fluoropentane	2.33	1
C <sub>4</sub> H <sub>4</sub> O	Furan	1.34	1	C <sub>5</sub> H <sub>11</sub> N	Piperidine	0.84	1
C <sub>4</sub> H <sub>4</sub> S	Thiophene	1.81	1	C <sub>5</sub> H <sub>11</sub> NO <sub>2</sub>	1-Nitropentane	2.01	1
C <sub>4</sub> H <sub>5</sub> N	Pyrrrole	0.75	1	C <sub>5</sub> H <sub>12</sub>	Pentane	3.45	1
C <sub>4</sub> H <sub>6</sub>	1,3-Butadiene	1.99	1	C <sub>5</sub> H <sub>12</sub>	Neopentane	3.11	1
C <sub>4</sub> H <sub>6</sub>	2-Butyne	1.46	1	C <sub>5</sub> H <sub>12</sub> O	1-Pentanol	1.51	1
C <sub>4</sub> H <sub>6</sub> O	2,5-Dihydrofuran	0.46	1	C <sub>5</sub> H <sub>12</sub> O	2-Pentanol	1.25	1
C <sub>4</sub> H <sub>6</sub> O <sub>2</sub>	Methacrylic acid	0.93	1	C <sub>5</sub> H <sub>12</sub> O	3-Pentanol	1.21	1
C <sub>4</sub> H <sub>6</sub> O <sub>2</sub>	Vinyl acetate	0.73	1	C <sub>5</sub> H <sub>12</sub> O	3-Methyl-1-butanol	1.28	1
C <sub>4</sub> H <sub>6</sub> O <sub>2</sub>	Methyl acrylate	0.80	1	C <sub>5</sub> H <sub>12</sub> O	2-Methyl-2-butanol	0.89	1
C <sub>4</sub> H <sub>7</sub> N	Butanenitrile	0.60	1	C <sub>5</sub> H <sub>12</sub> O	3-Methyl-2-butanol	1.28	1
C <sub>4</sub> H <sub>8</sub>	<i>cis</i> -2-Butene	2.33	1	C <sub>5</sub> H <sub>12</sub> O	2,2-Dimethyl-1-propanol	1.31	1
C <sub>4</sub> H <sub>8</sub>	<i>trans</i> -2-Butene	2.31	1	C <sub>5</sub> H <sub>12</sub> O	Methyl <i>tert</i> -butyl ether	0.94	1
C <sub>4</sub> H <sub>8</sub>	Isobutene	2.35	1	C <sub>5</sub> H <sub>13</sub> N	Pentylamine	1.49	1
C <sub>4</sub> H <sub>8</sub> Cl <sub>2</sub> O	Bis(2-chloroethyl) ether	1.12	2	C <sub>6</sub> Cl <sub>6</sub>	Hexachlorobenzene	5.47	5
C <sub>4</sub> H <sub>8</sub> O	Ethyl vinyl ether	1.04	1	C <sub>6</sub> HCl <sub>5</sub>	Pentachlorobenzene	5.03	5
C <sub>4</sub> H <sub>8</sub> O	Butanal	0.88	1	C <sub>6</sub> HCl <sub>5</sub> O	Pentachlorophenol	5.07	4
C <sub>4</sub> H <sub>8</sub> O	2-Butanone	0.29	1	C <sub>6</sub> H <sub>2</sub> Cl <sub>4</sub>	1,2,3,4-Tetrachlorobenzene	4.55	5
C <sub>4</sub> H <sub>8</sub> O	Tetrahydrofuran	0.46	1	C <sub>6</sub> H <sub>2</sub> Cl <sub>4</sub>	1,2,3,5-Tetrachlorobenzene	4.65	5
C <sub>4</sub> H <sub>8</sub> O <sub>2</sub>	Butanoic acid	0.79	1	C <sub>6</sub> H <sub>2</sub> Cl <sub>4</sub>	1,2,4,5-Tetrachlorobenzene	4.51	5
C <sub>4</sub> H <sub>8</sub> O <sub>2</sub>	Propyl formate	0.83	1	C <sub>6</sub> H <sub>3</sub> Cl <sub>3</sub>	1,2,3-Trichlorobenzene	4.04	5
C <sub>4</sub> H <sub>8</sub> O <sub>2</sub>	Ethyl acetate	0.73	1	C <sub>6</sub> H <sub>3</sub> Cl <sub>3</sub>	1,2,4-Trichlorobenzene	3.98	5
C <sub>4</sub> H <sub>9</sub> Br	1-Bromobutane	2.75	1	C <sub>6</sub> H <sub>3</sub> Cl <sub>3</sub>	1,3,5-Trichlorobenzene	4.02	5
C <sub>4</sub> H <sub>9</sub> Cl	1-Chlorobutane	2.64	2	C <sub>6</sub> H <sub>4</sub> Cl <sub>2</sub>	<i>o</i> -Dichlorobenzene	3.38	5
C <sub>4</sub> H <sub>9</sub> F	1-Fluorobutane	2.58	1	C <sub>6</sub> H <sub>4</sub> Cl <sub>2</sub>	<i>m</i> -Dichlorobenzene	3.48	5
C <sub>4</sub> H <sub>9</sub> I	1-Iodobutane	3	2	C <sub>6</sub> H <sub>4</sub> Cl <sub>2</sub>	<i>p</i> -Dichlorobenzene	3.38	5
C <sub>4</sub> H <sub>9</sub> N	Pyrrrolidine	0.46	1	C <sub>6</sub> H <sub>4</sub> Cl <sub>2</sub> O	2,4-Dichlorophenol	3.23	4
C <sub>4</sub> H <sub>9</sub> NO	Butanamide	-0.21	1	C <sub>6</sub> H <sub>5</sub> Br	Bromobenzene	2.99	2
C <sub>4</sub> H <sub>9</sub> NO	<i>N,N</i> -Dimethylacetamide	-0.77	1	C <sub>6</sub> H <sub>5</sub> Cl	Chlorobenzene	2.84	1
C <sub>4</sub> H <sub>9</sub> NO <sub>2</sub>	1-Nitrobutane	1.47	1	C <sub>6</sub> H <sub>5</sub> F	Fluorobenzene	2.27	2
C <sub>4</sub> H <sub>10</sub>	Isobutane	2.8	2	C <sub>6</sub> H <sub>5</sub> I	Iodobenzene	3.28	2
C <sub>4</sub> H <sub>10</sub> O	1-Butanol	0.84	1	C <sub>6</sub> H <sub>5</sub> NO <sub>2</sub>	Nitrobenzene	1.85	1
C <sub>4</sub> H <sub>10</sub> O	2-Butanol	0.65	1	C <sub>6</sub> H <sub>6</sub>	Benzene	2.13	1
C <sub>4</sub> H <sub>10</sub> O	2-Methyl-1-propanol	0.76	1	C <sub>6</sub> H <sub>6</sub> O	Phenol	1.48	4
C <sub>4</sub> H <sub>10</sub> O	2-Methyl-2-propanol	0.35	1	C <sub>6</sub> H <sub>6</sub> S	Benzenethiol	2.52	1
C <sub>4</sub> H <sub>10</sub> O	Diethyl ether	0.89	1	C <sub>6</sub> H <sub>7</sub> N	Aniline	0.90	1
C <sub>4</sub> H <sub>10</sub> S	1-Butanethiol	2.28	1	C <sub>6</sub> H <sub>7</sub> N	2-Methylpyridine	1.11	1
C <sub>4</sub> H <sub>10</sub> S	Diethyl sulfide	1.95	1	C <sub>6</sub> H <sub>7</sub> N	3-Methylpyridine	1.20	1
C <sub>4</sub> H <sub>11</sub> N	Butylamine	0.86	1	C <sub>6</sub> H <sub>7</sub> N	4-Methylpyridine	1.22	1
C <sub>4</sub> H <sub>11</sub> N	<i>tert</i> -Butylamine	0.40	1	C <sub>6</sub> H <sub>8</sub>	1,4-Cyclohexadiene	2.3	2
C <sub>4</sub> H <sub>11</sub> N	Diethylamine	0.58	1	C <sub>6</sub> H <sub>8</sub> O	5-Hexyn-2-one	0.58	1
C <sub>5</sub> H <sub>5</sub> N	Pyridine	0.65	1	C <sub>6</sub> H <sub>8</sub> O	2-Cyclohexen-1-one	0.61	1
C <sub>5</sub> H <sub>6</sub> O	2-Methylfuran	1.85	1	C <sub>6</sub> H <sub>8</sub> O	2-Ethylfuran	2.40	1
C <sub>5</sub> H <sub>7</sub> N	1-Methylpyrrrole	1.21	1	C <sub>6</sub> H <sub>10</sub>	1,5-Hexadiene	2.8	2
C <sub>5</sub> H <sub>8</sub>	1,4-Pentadiene	2.48	1	C <sub>6</sub> H <sub>10</sub>	1-Hexyne	2.73	2
C <sub>5</sub> H <sub>8</sub>	1-Pentyne	1.98	1	C <sub>6</sub> H <sub>10</sub>	Cyclohexene	2.86	1
C <sub>5</sub> H <sub>8</sub> O <sub>2</sub>	Methyl methacrylate	1.38	1	C <sub>6</sub> H <sub>10</sub> O	5-Hexen-2-one	1.02	1
C <sub>5</sub> H <sub>8</sub> O <sub>2</sub>	Ethyl acrylate	1.32	1	C <sub>6</sub> H <sub>10</sub> O	Cyclohexanone	0.81	1
C <sub>5</sub> H <sub>9</sub> N	Pentanenitrile	0.94	1	C <sub>6</sub> H <sub>10</sub> O <sub>2</sub>	Ethyl methacrylate	1.94	1
C <sub>5</sub> H <sub>10</sub>	1-Pentene	2.2	2	C <sub>6</sub> H <sub>11</sub> Br	Bromocyclohexane	3.20	1
C <sub>5</sub> H <sub>10</sub>	Cyclopentane	3.00	1	C <sub>6</sub> H <sub>11</sub> N	Hexanenitrile	1.66	1
C <sub>5</sub> H <sub>10</sub> O	2-Pentanone	0.84	1	C <sub>6</sub> H <sub>12</sub>	1-Hexene	3.40	1
C <sub>5</sub> H <sub>10</sub> O	3-Pentanone	0.82	1	C <sub>6</sub> H <sub>12</sub>	4-Methyl-1-pentene	2.5	2
C <sub>5</sub> H <sub>10</sub> O	3-Methyl-2-butanone	0.56	1	C <sub>6</sub> H <sub>12</sub>	Cyclohexane	3.44	1
C <sub>5</sub> H <sub>10</sub> O	Tetrahydropyran	0.82	1	C <sub>6</sub> H <sub>12</sub>	Methylcyclopentane	3.37	2

Mol. form.	Name	log <i>P</i>	Ref.	Mol. form.	Name	log <i>P</i>	Ref.
C <sub>6</sub> H <sub>12</sub> O	Cyclohexanol	1.23	1	C <sub>8</sub> H <sub>6</sub> O	Benzofuran	2.67	1
C <sub>6</sub> H <sub>12</sub> O	Hexanal	1.78	1	C <sub>8</sub> H <sub>6</sub> S	Benzo[b]thiophene	3.12	1
C <sub>6</sub> H <sub>12</sub> O	2-Hexanone	1.38	1	C <sub>8</sub> H <sub>7</sub> N	Benzeneacetonitrile	1.56	1
C <sub>6</sub> H <sub>12</sub> O	4-Methyl-2-pentanone	1.31	1	C <sub>8</sub> H <sub>7</sub> N	Indole	2.14	1
C <sub>6</sub> H <sub>12</sub> O <sub>2</sub>	Hexanoic acid	1.92	1	C <sub>8</sub> H <sub>8</sub>	Styrene	3.05	1
C <sub>6</sub> H <sub>12</sub> O <sub>2</sub>	Butyl acetate	1.82	1	C <sub>8</sub> H <sub>8</sub> O	Acetophenone	1.63	1
C <sub>6</sub> H <sub>13</sub> Br	1-Bromohexane	3.80	1	C <sub>8</sub> H <sub>8</sub> O	2-Methylbenzaldehyde	2.26	1
C <sub>6</sub> H <sub>13</sub> N	Cyclohexylamine	1.49	1	C <sub>8</sub> H <sub>8</sub> O	Benzeneacetaldehyde	1.78	1
C <sub>6</sub> H <sub>14</sub>	Hexane	4.00	1	C <sub>8</sub> H <sub>8</sub> O	2,3-Dihydrobenzofuran	2.14	1
C <sub>6</sub> H <sub>14</sub>	3-Methylpentane	3.60	2	C <sub>8</sub> H <sub>8</sub> O	Phenylloxirane	1.61	1
C <sub>6</sub> H <sub>14</sub>	2,2-Dimethylbutane	3.82	1	C <sub>8</sub> H <sub>8</sub> O <sub>2</sub>	<i>o</i> -Toluic acid	2.32	4
C <sub>6</sub> H <sub>14</sub>	2,3-Dimethylbutane	3.85	2	C <sub>8</sub> H <sub>8</sub> O <sub>2</sub>	<i>m</i> -Toluic acid	2.37	1
C <sub>6</sub> H <sub>14</sub> O	1-Hexanol	2.03	1	C <sub>8</sub> H <sub>8</sub> O <sub>2</sub>	<i>p</i> -Toluic acid	2.34	1
C <sub>6</sub> H <sub>14</sub> O	2-Hexanol	1.76	1	C <sub>8</sub> H <sub>8</sub> O <sub>2</sub>	Benzeneacetic acid	1.41	1
C <sub>6</sub> H <sub>14</sub> O	3-Hexanol	1.65	1	C <sub>8</sub> H <sub>8</sub> O <sub>2</sub>	Phenyl acetate	1.49	1
C <sub>6</sub> H <sub>14</sub> O	3,3-Dimethyl-2-butanol	1.48	1	C <sub>8</sub> H <sub>8</sub> O <sub>2</sub>	Methyl benzoate	2.20	1
C <sub>6</sub> H <sub>14</sub> O	Dipropyl ether	2.03	1	C <sub>8</sub> H <sub>10</sub>	Ethylbenzene	3.15	1
C <sub>6</sub> H <sub>14</sub> O	Diisopropyl ether	1.52	1	C <sub>8</sub> H <sub>10</sub>	<i>o</i> -Xylene	3.12	1
C <sub>6</sub> H <sub>15</sub> N	Hexylamine	2.06	1	C <sub>8</sub> H <sub>10</sub>	<i>m</i> -Xylene	3.20	1
C <sub>6</sub> H <sub>15</sub> N	Dipropylamine	1.67	1	C <sub>8</sub> H <sub>10</sub>	<i>p</i> -Xylene	3.15	1
C <sub>6</sub> H <sub>15</sub> N	Triethylamine	1.45	1	C <sub>8</sub> H <sub>10</sub> O	<i>o</i> -Ethylphenol	2.47	1
C <sub>7</sub> H <sub>5</sub> BrO <sub>2</sub>	2-Bromobenzoic acid	2.20	4	C <sub>8</sub> H <sub>10</sub> O	<i>m</i> -Ethylphenol	2.50	1
C <sub>7</sub> H <sub>5</sub> BrO <sub>2</sub>	3-Bromobenzoic acid	2.87	4	C <sub>8</sub> H <sub>10</sub> O	<i>p</i> -Ethylphenol	2.50	1
C <sub>7</sub> H <sub>5</sub> BrO <sub>2</sub>	4-Bromobenzoic acid	2.86	4	C <sub>8</sub> H <sub>10</sub> O	2,4-Xylenol	2.35	1
C <sub>7</sub> H <sub>5</sub> N	Benzonitrile	1.56	1	C <sub>8</sub> H <sub>10</sub> O	2,5-Xylenol	2.34	1
C <sub>7</sub> H <sub>6</sub> O	Benzaldehyde	1.48	1	C <sub>8</sub> H <sub>10</sub> O	2,6-Xylenol	2.36	1
C <sub>7</sub> H <sub>6</sub> O <sub>2</sub>	Benzoic acid	1.88	4	C <sub>8</sub> H <sub>10</sub> O	3,4-Xylenol	3.23	1
C <sub>7</sub> H <sub>6</sub> O <sub>2</sub>	Phenyl formate	1.26	1	C <sub>8</sub> H <sub>10</sub> O	3,5-Xylenol	2.35	1
C <sub>7</sub> H <sub>6</sub> O <sub>3</sub>	Salicylic acid	2.20	4	C <sub>8</sub> H <sub>10</sub> O	Benzeneethanol	1.36	1
C <sub>7</sub> H <sub>7</sub> Br	(Bromomethyl)benzene	2.92	1	C <sub>8</sub> H <sub>10</sub> O	$\alpha$ -Methylbenzyl alcohol	1.42	1
C <sub>7</sub> H <sub>7</sub> Cl	<i>o</i> -Chlorotoluene	3.42	1	C <sub>8</sub> H <sub>10</sub> O	3-Methylbenzenemethanol	1.60	1
C <sub>7</sub> H <sub>7</sub> Cl	<i>m</i> -Chlorotoluene	3.28	1	C <sub>8</sub> H <sub>10</sub> O	4-Methylbenzenemethanol	1.58	1
C <sub>7</sub> H <sub>7</sub> Cl	<i>p</i> -Chlorotoluene	3.33	1	C <sub>8</sub> H <sub>10</sub> O	Phenetole	2.51	1
C <sub>7</sub> H <sub>7</sub> Cl	(Chloromethyl)benzene	2.30	1	C <sub>8</sub> H <sub>10</sub> O	Benzyl methyl ether	1.35	1
C <sub>7</sub> H <sub>7</sub> NO <sub>2</sub>	<i>p</i> -Nitrotoluene	2.42	1	C <sub>8</sub> H <sub>10</sub> O	2-Methylanisole	2.74	1
C <sub>7</sub> H <sub>8</sub>	Toluene	2.73	1	C <sub>8</sub> H <sub>10</sub> O	3-Methylanisole	2.66	1
C <sub>7</sub> H <sub>8</sub>	1,3,5-Cycloheptatriene	2.63	2	C <sub>8</sub> H <sub>10</sub> O	4-Methylanisole	2.81	1
C <sub>7</sub> H <sub>8</sub> O	<i>o</i> -Cresol	1.98	1	C <sub>8</sub> H <sub>11</sub> N	<i>p</i> -Ethylaniline	1.96	1
C <sub>7</sub> H <sub>8</sub> O	<i>m</i> -Cresol	1.98	1	C <sub>8</sub> H <sub>11</sub> N	<i>N,N</i> -Dimethylaniline	2.31	1
C <sub>7</sub> H <sub>8</sub> O	<i>p</i> -Cresol	1.97	1	C <sub>8</sub> H <sub>11</sub> N	Benzeneethanamine	1.41	1
C <sub>7</sub> H <sub>8</sub> O	Benzyl alcohol	1.05	1	C <sub>8</sub> H <sub>14</sub> O <sub>2</sub>	Butyl methacrylate	2.88	1
C <sub>7</sub> H <sub>8</sub> O	Anisole	2.11	1	C <sub>8</sub> H <sub>15</sub> N	Octanenitrile	2.75	1
C <sub>7</sub> H <sub>9</sub> N	Benzylamine	1.09	1	C <sub>8</sub> H <sub>16</sub>	1-Octene	4.57	1
C <sub>7</sub> H <sub>9</sub> N	<i>o</i> -Methylaniline	1.32	1	C <sub>8</sub> H <sub>16</sub>	Cyclooctane	4.45	2
C <sub>7</sub> H <sub>9</sub> N	<i>m</i> -Methylaniline	1.40	1	C <sub>8</sub> H <sub>16</sub> O	2-Octanone	2.37	1
C <sub>7</sub> H <sub>9</sub> N	<i>p</i> -Methylaniline	1.39	1	C <sub>8</sub> H <sub>16</sub> O <sub>2</sub>	Octanoic acid	3.05	1
C <sub>7</sub> H <sub>9</sub> N	<i>N</i> -Methylaniline	1.66	1	C <sub>8</sub> H <sub>17</sub> Br	1-Bromooctane	4.89	1
C <sub>7</sub> H <sub>14</sub>	1-Heptene	3.99	1	C <sub>8</sub> H <sub>18</sub>	Octane	5.15	1
C <sub>7</sub> H <sub>14</sub>	Methylcyclohexane	3.88	1	C <sub>8</sub> H <sub>18</sub> O	1-Octanol	3.07	1
C <sub>7</sub> H <sub>14</sub> O	2-Heptanone	1.98	1	C <sub>8</sub> H <sub>18</sub> O	2-Octanol	2.90	1
C <sub>7</sub> H <sub>14</sub> O	5-Methyl-2-hexanone	1.88	1	C <sub>8</sub> H <sub>18</sub> O	4-Octanol	2.68	1
C <sub>7</sub> H <sub>15</sub> Br	1-Bromoheptane	4.36	1	C <sub>8</sub> H <sub>18</sub> O	Dibutyl ether	3.21	1
C <sub>7</sub> H <sub>15</sub> Cl	1-Chloroheptane	4.15	1	C <sub>9</sub> H <sub>7</sub> N	Quinoline	2.03	1
C <sub>7</sub> H <sub>15</sub> I	1-Iodoheptane	4.70	1	C <sub>9</sub> H <sub>7</sub> N	Isoquinoline	2.08	1
C <sub>7</sub> H <sub>16</sub>	Heptane	4.50	1	C <sub>9</sub> H <sub>8</sub>	Indene	2.92	1
C <sub>7</sub> H <sub>16</sub> O	1-Heptanol	2.62	1	C <sub>9</sub> H <sub>8</sub> O <sub>2</sub>	<i>trans</i> -Cinnamic acid	2.13	1
C <sub>7</sub> H <sub>16</sub> O	2-Heptanol	2.31	1	C <sub>9</sub> H <sub>9</sub> N	Benzenepropanenitrile	1.72	1
C <sub>7</sub> H <sub>16</sub> O	3-Heptanol	2.24	1	C <sub>9</sub> H <sub>10</sub>	Indan	3.33	1
C <sub>7</sub> H <sub>16</sub> O	4-Heptanol	2.22	1	C <sub>9</sub> H <sub>10</sub> O	1-Phenyl-1-propanone	2.19	1
C <sub>7</sub> H <sub>17</sub> N	Heptylamine	2.57	1	C <sub>9</sub> H <sub>10</sub> O	1-Phenyl-2-propanone	1.44	1
C <sub>8</sub> H <sub>6</sub>	Phenylacetylene	2.40	1	C <sub>9</sub> H <sub>10</sub> O	4-Methylacetophenone	2.19	1

Mol. form.	Name	log <i>P</i>	Ref.	Mol. form.	Name	log <i>P</i>	Ref.
C <sub>9</sub> H <sub>10</sub> O <sub>2</sub>	2-Phenylpropanoic acid	1.80	1	C <sub>12</sub> H <sub>6</sub> Cl <sub>4</sub>	2,2',4',5-Tetrachlorobiphenyl	5.73	7
C <sub>9</sub> H <sub>10</sub> O <sub>2</sub>	Benzyl acetate	1.96	1	C <sub>12</sub> H <sub>7</sub> Cl <sub>3</sub>	2,4,5-Trichlorobiphenyl	5.60	3
C <sub>9</sub> H <sub>10</sub> O <sub>2</sub>	4-Methylphenyl acetate	2.11	1	C <sub>12</sub> H <sub>7</sub> Cl <sub>3</sub>	2,4,6-Trichlorobiphenyl	5.47	3
C <sub>9</sub> H <sub>10</sub> O <sub>2</sub>	Ethyl benzoate	2.64	1	C <sub>12</sub> H <sub>8</sub> Cl <sub>2</sub>	2,5-Dichlorobiphenyl	5.10	3
C <sub>9</sub> H <sub>12</sub>	Propylbenzene	3.69	1	C <sub>12</sub> H <sub>8</sub> Cl <sub>2</sub>	2,6-Dichlorobiphenyl	5.00	3
C <sub>9</sub> H <sub>12</sub>	Isopropylbenzene	3.66	1	C <sub>12</sub> H <sub>8</sub> O	Dibenzofuran	4.12	1
C <sub>9</sub> H <sub>12</sub>	<i>o</i> -Ethyltoluene	3.53	1	C <sub>12</sub> H <sub>9</sub> Cl	2-Chlorobiphenyl	4.52	1
C <sub>9</sub> H <sub>12</sub>	<i>p</i> -Ethyltoluene	3.63	2	C <sub>12</sub> H <sub>9</sub> Cl	3-Chlorobiphenyl	4.58	1
C <sub>9</sub> H <sub>12</sub>	1,2,3-Trimethylbenzene	3.60	1	C <sub>12</sub> H <sub>9</sub> Cl	4-Chlorobiphenyl	4.61	1
C <sub>9</sub> H <sub>12</sub>	1,2,4-Trimethylbenzene	3.63	1	C <sub>12</sub> H <sub>9</sub> N	Carbazole	3.72	1
C <sub>9</sub> H <sub>12</sub>	1,3,5-Trimethylbenzene	3.42	1	C <sub>12</sub> H <sub>10</sub>	Acenaphthene	3.96	4
C <sub>9</sub> H <sub>12</sub> O	2-Propylphenol	2.93	1	C <sub>12</sub> H <sub>10</sub>	Biphenyl	3.76	6
C <sub>9</sub> H <sub>12</sub> O	4-Propylphenol	3.20	1	C <sub>12</sub> H <sub>10</sub> N <sub>2</sub>	Azobenzene	3.82	1
C <sub>9</sub> H <sub>12</sub> O	2,3,6-Trimethylphenol	2.67	1	C <sub>12</sub> H <sub>10</sub> O	Diphenyl ether	4.21	1
C <sub>9</sub> H <sub>12</sub> O	2,4,6-Trimethylphenol	2.46	1	C <sub>12</sub> H <sub>10</sub> S	Diphenyl sulfide	4.45	1
C <sub>9</sub> H <sub>12</sub> O	Benzenepropanol	1.88	1	C <sub>12</sub> H <sub>11</sub> N	Diphenylamine	3.44	4
C <sub>9</sub> H <sub>13</sub> N	<i>N,N</i> -Dimethylbenzylamine	1.98	1	C <sub>12</sub> H <sub>12</sub>	1-Ethyl-naphthalene	4.40	1
C <sub>9</sub> H <sub>13</sub> N	Amphetamine	1.76	1	C <sub>12</sub> H <sub>12</sub>	1,2-Dimethylnaphthalene	4.31	1
C <sub>9</sub> H <sub>18</sub>	1-Nonene	5.15	1	C <sub>12</sub> H <sub>12</sub>	1,4-Dimethylnaphthalene	4.37	1
C <sub>9</sub> H <sub>18</sub> O	2-Nonanone	3.16	1	C <sub>12</sub> H <sub>14</sub> O	4-Phenylcyclohexanone	2.45	1
C <sub>9</sub> H <sub>18</sub> O	5-Methyl-2-octanone	2.92	1	C <sub>12</sub> H <sub>18</sub>	Hexylbenzene	5.52	1
C <sub>9</sub> H <sub>20</sub>	Nonane	5.65	1	C <sub>12</sub> H <sub>18</sub>	Hexamethylbenzene	4.69	4
C <sub>9</sub> H <sub>20</sub> O	1-Nonanol	4.02	1	C <sub>12</sub> H <sub>22</sub> O	Cyclododecanone	4.10	1
C <sub>9</sub> H <sub>21</sub> N	Tripropylamine	2.79	1	C <sub>12</sub> H <sub>24</sub> O <sub>2</sub>	Dodecanoic acid	4.6	1
C <sub>10</sub> H <sub>7</sub> Cl	1-Chloronaphthalene	3.90	1	C <sub>12</sub> H <sub>26</sub> O	1-Dodecanol	5.13	1
C <sub>10</sub> H <sub>7</sub> Cl	2-Chloronaphthalene	3.98	1	C <sub>13</sub> H <sub>8</sub> O	9H-Fluorene-9-one	3.58	1
C <sub>10</sub> H <sub>8</sub>	Naphthalene	3.34	4	C <sub>13</sub> H <sub>9</sub> N	Acridine	3.40	1
C <sub>10</sub> H <sub>8</sub>	Azulene	3.22	1	C <sub>13</sub> H <sub>10</sub>	9H-Fluorene	4.20	4
C <sub>10</sub> H <sub>8</sub> O	1-Naphthol	2.84	1	C <sub>13</sub> H <sub>10</sub> O	Benzophenone	3.18	1
C <sub>10</sub> H <sub>8</sub> O	2-Naphthol	2.70	1	C <sub>13</sub> H <sub>10</sub> O <sub>2</sub>	Phenyl benzoate	3.59	1
C <sub>10</sub> H <sub>12</sub> O <sub>2</sub>	Isopropyl benzoate	3.18	1	C <sub>13</sub> H <sub>11</sub> NO	<i>N</i> -Phenylbenzamide	2.62	1
C <sub>10</sub> H <sub>14</sub>	Butylbenzene	4.26	1	C <sub>13</sub> H <sub>12</sub>	Diphenylmethane	4.14	1
C <sub>10</sub> H <sub>14</sub>	<i>tert</i> -Butylbenzene	4.11	1	C <sub>13</sub> H <sub>12</sub>	4-Methylbiphenyl	4.63	1
C <sub>10</sub> H <sub>14</sub>	Isobutylbenzene	4.01	2	C <sub>13</sub> H <sub>12</sub> O	Diphenylmethanol	2.67	1
C <sub>10</sub> H <sub>14</sub>	<i>p</i> -Cymene	4.10	1	C <sub>13</sub> H <sub>12</sub> O	Benzyl phenyl ether	3.79	1
C <sub>10</sub> H <sub>14</sub>	1,2,4,5-Tetramethylbenzene	4.10	2	C <sub>14</sub> H <sub>10</sub>	Anthracene	4.56	4
C <sub>10</sub> H <sub>14</sub>	1,2,3,4-Tetramethylbenzene	4.00	1	C <sub>14</sub> H <sub>10</sub>	Phenanthrene	4.52	4
C <sub>10</sub> H <sub>14</sub>	1,2,3,5-Tetramethylbenzene	4.10	1	C <sub>14</sub> H <sub>12</sub>	<i>trans</i> -Stilbene	4.81	1
C <sub>10</sub> H <sub>14</sub> O	4-Butylphenol	3.65	1	C <sub>14</sub> H <sub>12</sub>	1-Methylfluorene	4.97	1
C <sub>10</sub> H <sub>20</sub> O	2-Decanone	3.77	1	C <sub>14</sub> H <sub>12</sub> O	2-Phenylacetophenone	3.18	1
C <sub>10</sub> H <sub>20</sub> O <sub>2</sub>	Decanoic acid	4.09	1	C <sub>14</sub> H <sub>12</sub> O <sub>2</sub>	Benzyl benzoate	3.97	1
C <sub>10</sub> H <sub>22</sub>	Decane	6.25	1	C <sub>14</sub> H <sub>14</sub>	1,2-Diphenylethane	4.70	1
C <sub>10</sub> H <sub>22</sub> O	1-Decanol	4.57	1	C <sub>14</sub> H <sub>14</sub>	4,4'-Dimethylbiphenyl	5.09	1
C <sub>11</sub> H <sub>9</sub> N	4-Phenylpyridine	2.59	1	C <sub>14</sub> H <sub>22</sub>	Octylbenzene	6.30	1
C <sub>11</sub> H <sub>10</sub>	1-Methylnaphthalene	3.87	1	C <sub>14</sub> H <sub>28</sub> O <sub>2</sub>	Tetradecanoic acid	6.1	1
C <sub>11</sub> H <sub>10</sub>	2-Methylnaphthalene	4.00	1	C <sub>15</sub> H <sub>12</sub>	2-Methylanthracene	5.15	2
C <sub>11</sub> H <sub>16</sub>	Pentylbenzene	4.90	1	C <sub>15</sub> H <sub>12</sub>	9-Methylanthracene	5.07	1
C <sub>11</sub> H <sub>16</sub>	Pentamethylbenzene	4.56	1	C <sub>15</sub> H <sub>12</sub>	1-Methylphenanthrene	5.14	2
C <sub>11</sub> H <sub>22</sub> O	2-Undecanone	4.09	1	C <sub>16</sub> H <sub>10</sub>	Fluoranthene	5.07	4
C <sub>11</sub> H <sub>22</sub> O <sub>2</sub>	Methyl decanoate	4.41	1	C <sub>16</sub> H <sub>10</sub>	Pyrene	5.08	4
C <sub>12</sub> Cl <sub>10</sub>	Decachlorobiphenyl	8.26	3	C <sub>16</sub> H <sub>14</sub>	9,10-Dimethylanthracene	5.69	1
C <sub>12</sub> HCl <sub>9</sub>	2,2',3,3',4,5,5',6,6'- Nonachlorobiphenyl	8.16	3	C <sub>16</sub> H <sub>32</sub> O <sub>2</sub>	Hexadecanoic acid	7.17	1
C <sub>12</sub> H <sub>2</sub> Cl <sub>8</sub>	2,2',3,3',5,5',6,6'- Octachlorobiphenyl	7.10	3	C <sub>17</sub> H <sub>12</sub>	11H- <i>Benzo</i> [a]fluorene	5.40	1
C <sub>12</sub> H <sub>3</sub> Cl <sub>7</sub>	2,2',3,3',4,4',6-Heptachlorobiphenyl	6.70	3	C <sub>17</sub> H <sub>12</sub>	11H- <i>Benzo</i> [b]fluorene	5.75	1
C <sub>12</sub> H <sub>4</sub> Cl <sub>6</sub>	2,2',3,3',4,4'-Hexachlorobiphenyl	7.00	3	C <sub>18</sub> H <sub>12</sub>	Benz[a]anthracene	5.91	1
C <sub>12</sub> H <sub>4</sub> Cl <sub>6</sub>	2,2',4,4',6,6'-Hexachlorobiphenyl	7.00	3	C <sub>18</sub> H <sub>12</sub>	Chrysene	5.73	4
C <sub>12</sub> H <sub>4</sub> Cl <sub>6</sub>	2,2',3,3',6,6'-Hexachlorobiphenyl	6.70	3	C <sub>18</sub> H <sub>12</sub>	Naphthacene	5.76	1
C <sub>12</sub> H <sub>5</sub> Cl <sub>5</sub>	2,3,4,5,6-Pentachlorobiphenyl	6.30	3	C <sub>18</sub> H <sub>12</sub>	Triphenylene	5.49	4
C <sub>12</sub> H <sub>5</sub> Cl <sub>5</sub>	2,2',4,5,5'-Pentachlorobiphenyl	6.40	3	C <sub>18</sub> H <sub>15</sub> N	Triphenylamine	5.74	1
C <sub>12</sub> H <sub>6</sub> Cl <sub>4</sub>	2,3,4,5-Tetrachlorobiphenyl	5.72	3	C <sub>18</sub> H <sub>30</sub> O <sub>2</sub>	Linolenic acid	6.46	1
				C <sub>18</sub> H <sub>32</sub> O <sub>2</sub>	Linoleic acid	7.05	1

Mol. form.	Name	log <i>P</i>	Ref.	Mol. form.	Name	log <i>P</i>	Ref.
C <sub>18</sub> H <sub>34</sub> O <sub>2</sub>	Oleic acid	7.64	1	C <sub>20</sub> H <sub>40</sub> O <sub>2</sub>	Arachidic acid	9.29	1
C <sub>18</sub> H <sub>36</sub> O <sub>2</sub>	Stearic acid	8.23	1	C <sub>21</sub> H <sub>16</sub>	1,2-Dihydro-3-methylbenz[j] aceanthrylene	6.75	1
C <sub>19</sub> H <sub>16</sub> O	Triphenylmethanol	3.68	1	C <sub>22</sub> H <sub>12</sub>	Benzo[ghi]perylene	6.90	1
C <sub>20</sub> H <sub>12</sub>	Perylene	6.25	1	C <sub>24</sub> H <sub>12</sub>	Coronene	6.05	4
C <sub>20</sub> H <sub>12</sub>	Benzo[a]pyrene	6.20	4				
C <sub>20</sub> H <sub>32</sub> O <sub>2</sub>	Arachidonic acid	6.98	1				